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Exam. Code : 103201 Subject Code : 1269

# B.A./B.Sc. 1<sup>st</sup> Semester CHEMISTRY (Organic Chemistry—I)

Time Allowed—Three Hours] [Maximum Marks—35

Note :— Attempt FIVE questions selecting at least ONE question from each section. The fifth question may be attempted from any section. All questions carry equal marks.

#### SECTION-A

- (a) Out of ethylamine and acetamide, which one is more basic and why ?
   3
  - (b) An aqueous solution of tropyllium bromide on treatment with AgNO<sub>3</sub> gives precipitate of AgBr. Explain.
- (a) Cyclohexnone exists in keto form while cyclohexa-2,
  4-dien-1-one exists in enol form, explain. 3
  - (b) Arrange the following carbanions in increasing order of stability and justify :

4

(Contd.)

 $^{\odot}$ CH<sub>2</sub>COOC<sub>2</sub>H<sub>5</sub>, (CH<sub>3</sub>)<sub>3</sub>C $^{\ominus}$ , (CH<sub>3</sub>)<sub>2</sub>HC $^{\ominus}$ 

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### SECTION-B

- (a) Discuss Corey-House reaction with a suitable example.
  3
  - (b) Complete the following reaction and provide a suitable mechanism :



4. (a) Complete the following reaction and provide a suitable mechanism : 3



(b) Complete the following reaction by providing the structure of A and B. Also provide a suitable mechanism for the formation of B : 4

$$\xrightarrow{\text{Na}} A \xrightarrow{\text{Na/liq. NH}_3} B$$

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### SECTION—C

- 5. (a) Enlist the differences between  $S_N 1$  and  $S_N 2$ reaction. 5
  - (b) How do you account for the racemization of (+)-2-iodobutane when it is allowed to react with iodide ion ?
- Discuss Baeyer's strain theory. How it is used to explain the reactivity of cyclopropane and cyclobutane rings ? Also discuss its limitations.

### SECTION-D

7. (a) Complete the following reaction and provide a suitable mechanism : 5



(b) Predict the product of mono-nitration of the following : 2





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3

(Contd.)

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8. Predict the product/products in the following reactions :

3+2+2



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